

Preparation and Characterization of Lornoxicam- B-Cyclodextrin Inclusion Complex for Solubility Enhancement & Taste Masking

Saurabh Sharma¹* and Anurag Bhargav²

1. Research Scholar, CMJ University, Shillong, Meghalaya, India -793 003

2. Chaudhari Devilal College of Pharmacy, Yamunanagar, Haryana, India

Abstract

Lornoxicam (LXM) is a practically insoluble BCS Class II drug. Attempt is made to convert an insoluble drug into a water soluble complex by solubility enhancement techniques like by kneading method, spray drying method and complexation method using β -Cyclodextrin. The Complex was characterized by Differential Scanning Calorimetry (DSC), Powder X-ray Diffraction (XRD), Fourier-transform infrared spectroscopy (FT-IR), Solubility and Dissolution studies. According to the DSC data, lornoxicam in complex form was converted from crystalline to amorphous state resulting in better dissolution rate. FTIR study reveals that there is no interaction between drug and polymer. The PXRD study revealed that the crystallinity of lornoxicam is reduced to amorphous. The dissolution study was conducted in dissolution medium having pH 6.8.. The solid dispersion lornoxicam-β-CD inclusion complex showed of maximum 99.67 \pm 0.75% drug release in 90 min than physical mixture $41.10 \pm 0.50\%$ drug release in 120 min and Pure drug lornoxicam only $37.50 \pm 0.82\%$ release in 120 min., resulting in enhanced dissolution rate and bioavailability. It can be concluded that inclusion complex of lornoxicam with β -CD was prepared by kneading method used for taste masking of bitter drug (lornoxicam) and solubility enhancement were found to be effective for solubility and dissolution rate leads to enhanced bioavailability of lornoxicam.

Key Words: Lornoxicam; β -Cyclodextrin; kneading method; spray drying method; complexation method; Solubility; Dissolution rate.

Introduction

Poorly water-soluble drugs often show low bioavailability when administered orally, because the absorption of the drugs in the gastro-intestinal tract can usually be a ratelimiting step. Therefore, it is important for such kind of drugs to enhance their dissolution rate. To enhance the dissolution rate, increasing the drug solubility is necessary according to the Noyes–Whitney equation¹. Various studies have been done in attempt to improve solubilities of poorly water-soluble drugs; they include micronization², solid dispersion³, solvent deposition⁴, ordered mixture⁵, roll-mixing⁶, complexation⁷, cogrinding method⁸, microparticles⁹⁻¹⁰, nanoparticles¹⁰⁻¹¹,

* Corresponding Author

E.mail: saurabhapex1975@gmail.com

liquisolid compacts¹², ball milled¹³, dry mixing¹⁴, grinding¹⁵, cosolvent¹⁶. Lornoxicam is a non-steroidal anti-inflammatory drug with extremely potent anti-inflammatory and analgesic activity. It is widely used for the symptomatic treatment of pain and inflammation in patients with rheumatoid arthritis and osteoarthritis. Moreover it showed great efficacy in various clinical trials in the management of preoperative and postoperative pain associated with gynecological, orthopedic, abdominal and dental surgeries. Lornoxicam shows bitter taste and distinct pH-dependent solubility characterized by very poor solubility in acidic condition present in the stomach¹⁷.

Cyclodextrins (CDs) are known for their ability to molecular encapsulates a wide variety of drugs into their hydrophobic cavity¹⁸. The interaction of CDs with labile compounds can retard drug degradation, accelerate degradation, or have no effect on drug molecules reactivity¹⁹⁻²⁰. The main purpose of this study is to enhance the solubility, increase the dissolution rate and bioavailability of Lornoxicam which is well demonstrated by Differential scanning calorimetry (DSC,) X-ray powder diffractometry (XRD), Fourier transform infrared spectroscopy (FTIR), solubility study and dissolution study. The aim of present work is to uplift the solubility and dissolution rate leads to enhance the bioavailability of lornoxicam.

MATERIAL AND METHOD

Lornoxicam was gratis sample from Piramal Healthcare Pvt. Ltd., Mumbai. India. β - Cyclodextrin was purchased from HI-MEDIA. All other materials and reagents were of analytical grade of purity.

Preparation of Solid Dispersion

Ratio optimization of drug and β-Cyclodextrin

Apparent solubility method has been used for ratio optimization. Physical mixtures of drug- β -Cyclodextrin (β -CD) in different ratios were prepared²¹. Then Physical mixtures equivalent to (10 mg of drug was added to 10 mL of 0.1N HCl having pH 6.8) i.e. 1:1 similarly 1:2, 1:3 & 1:4 were taken in glass vials with rubber closures. Then the vials were kept on a shaker incubator maintained at 37±0.5°C for 24 h. After shaking, the vials were kept in an incubator at 37±0.5°C for equilibrium for 12 h. The solution was then filtered through 0.45 µm millipore filter and the filtrate was assayed spectrophotometrically at λ_{max} 377.5 nm. From the results shown in (Table 1 and Fig.1) the drug- β -CD ratio was optimized and used for solid dispersions formulation.

Preparation of drug-β-CD inclusion complex by kneading method

Solid dispersion with a optimized ratio of Lornoxicam with β -CD were prepared with Kneading method. In this a mixture of Lornoxicam- β -CD Complex (1:2) were mixed in mortar with a pestle and the required amount of solvent (methanol–water 1:1 mixture) just to make a smooth paste was added. The paste was then further kneaded for about 1 h in a glass mortar. A similar method was reported by Fernandes et al²². The paste formed was dried under vacuum for 24 hours. Dried powder was passed through sieve no. 100 and stored in a desiccator until further evaluation See (Table 2 & Fig.2)

Approximately 10 L of an aqueous solution containing lornoxicam/β-cyclodextrin complex were prepared following the above-described phase-solubility study procedure, at a molar ratio of 1:2 (6 mM of lornoxicam and 12 mM of β-cyclodextrin). The solution was spraydried using a Niro Production spray dryer, under the following operating conditions: inlet air temperature, 175°C; outlet air temperature, 99°C; atomizer rotation rate, 10,900 rpm; and feed solution flow, 143 mL min⁻¹. The yield of the spray-drying process was measured as the powder weight percentage obtained at the end of the operation compared with the amount of solid materials (lornoxicam/β-cyclodextrin) present in the sprayed solution.²³ Dried powder was passed through sieve no. 100 and stored in a desiccator until further evaluation See (Table 2 & Fig.2)

In this method complex was prepared in ratio of 1:2 w/w completely dissolving lornoxicam in organic solvent (methanol) and finally β -CD was added with continuous stirring until homogenous mixture is formed. The mixture

was then dried in an oven for 6 hrs at 45 ⁰C.²¹ Dried powder was passed through sieve no. 100 and stored in a desiccator until further evaluation See (Table 2 & Fig.2)

Evaluation of Solid dispersions

Saturation solubility studies

Saturation solubility studies were performed according to

the method reported by Higuchi and Connors to analyze the improvement in solubility in 0.1N HCl having pH 1.2 in triplicate. Excess of pure drug, physical mixture, and inclusion complex were added to 20ml of distilled water in glass vials which were subsequently tightly closed and shaken for 24 h on a mechanical shaker at room temperature to achieve the equilibrium. In preliminary studies, it was found that equilibrium solubility was achieved in 24 h and therefore, samples were shaken for 24 h. The samples were withdrawn and filtered through a 0.22 μ m membrane filter (Millipore, India) followed by dilution and analyzed lornoxicam content was determined in triplicate by (UV/Vis Spectrophotometer, Shimadzu-1800) spectrophotometrically at at λ max 377.5 nm. The results were shown in (Table 2 & Fig.1)

Production Yield

The yields of production of solid dispersions of various batches were calculated using the weight of final product after drying with respect to the initial total weight of the drug and polymer used for preparation of solid dispersions and percent production yields were calculated as per the formula mentioned below,²⁶ and are reported in (Table 3 & Fig.2)

	Practical mass (solid dispersions)				
% yield = -		Х	100		
	Theoretical mass (polymer + drug)				

Drug Content

Solid dispersions equivalent to 10 mg of drug according to theoretical mass was dissolved in the sufficient amount of methanol into 100 mL volumetric flask and volume was made up to the mark by methanol.²⁶ Then the concentration of drug present in the solution was determined using UV spectroscopy by taking absorbance at λ_{max} 377.5 nm. Results are reported in (Table 3 & Fig.3)

Taste Evaluation of Lornoxicam with $\beta\text{-}CD$ inclusion complex

Taste evaluation of lornoxicam with β -CD inclusion complex with ratio 1:2 prepared by kneading method was done to compare bitterness of the optimized solid dispersion inclusion complex with pure drugs. Time sensitivity method was used for bitterness evaluation, in which 10 healthy human volunteers were selected and a sample of solid dispersion inclusion complex equivalents to 10mg (oral dose) were kept on tongue in mouth cavity without swallowing for 10-15sec and all the volunteers were asked for verbal judgment followed by scoring in a taste scale of 0-3, finally all the volunteers were asked to gargle their mouth by distilled water after competition of taste evaluation. Bitterness levels were noted by verbal judgments at different time intervals.²⁷ Results were shown in (Table 4)

Taste scale was calibrated as follows:

- 0 Being = Tasteless
- X Being= Thresholds bitter
- 1 Being= Slight bitter
- 2 Being= Moderate bitter
- 3 Being= Strong bitter

Characterization of Solid Dispersions Fourier Transform Infra red Spectroscopy (FTIR)

The optimized lornoxicam with β -CD inclusion complex with ratio 1:2 prepared by kneading method were subjected to fourier transform infra red spectroscopy (FTIR) studies to check drug interaction and complex

http://www.ijddhrjournal.com.

formation using FTIR (IR IFFINITY-1 CE, Shimadzu corps, Japan). The sample of pure drug, β -CD, physical mixture and solid dispersion inclusion complex were previously grounded and thoroughly mixed with KBr and formed an infrared transparent solid disk. The KBr disks were prepared by compressing the powder blend. Three scans were executed at a resolution of 1 cm⁻¹ (from 4000-

400 cm⁻¹). Results are reported in (Fig.4)

Differential Scanning Calorimetry (DSC)

DSC curves of lornoxicam, β -CD, physical mixtures solid dispersions inclusion complex were obtained by a differential scanning calorimeter (DSC 60 Shimadzu) at a heating rate of 10 °C/min from 30 to 300 °C in nitrogen atmosphere, which are shown in (Fig.5)

X-Ray Powder Diffraction Studies (XRPD)

Powder XRD patterns of lornoxicam, β -CD physical mixtures solid dispersions inclusion complex were recorded using Philips diffractometer (PW 1140) and Cu-ka radiation, diffractograms were run at a scanning speed of 2°/mm and a chart speed of 2°/2 cm per 2ø, which are shown in (Fig.6)

Scanning Electron Microscopy (SEM)

The surface morphology of pure drug, physical mixture and solid dispersion inclusion complexes with β -CD prepared by various methods was examined by Scanning Electron Microscope (JEOL 5400, Tokyo, Japan). In this study the samples were fixed on a brass stub using double-sided tape and then gold coated in vacuum by a sputter coater. The pictures were then taken at an excitation voltage of 15 kV. Results are reported in (Fig.7 & 8)

In-vitro Dissolution Study

The in-vitro dissolution of lornoxicam, physical mixture and prepared solid inclusion complexes with B-CD and was carried out using powder dispersion method by using Dissolution test apparatus-TDT-06T (Electrolab, Mumbai, India) at the USP type II apparatus at 100 rpm. The dissolution study was conducted in dissolution medium of having pH 6.8 as a dissolution media at $37^{\circ}C + 0.5^{\circ}C$. Optimized batches of ratio 1:2 lornoxicam-β-CD inclusion complex containing equivalent of 10mg of lornoxicam were suspended in 900 ml of pH 6.8 and A 5 ml aliquot of dissolution medium was withdrawn at 5, 10, 15, 30, 45, 60, 90, 105, 120 min with a pipette and filter through 0.45 µm Whatman filter and then analyzed lornoxicam content was determined in triplicate by Spectrophotometer, Shimadzu-1800) (UV/Vis spectrophotometrically at at λmax 377.5 nm. Fresh medium (5 ml), which was prewarmed at 37°C, was replaced immediately into the dissolution medium after each sampling maintain its constant volume throughout the test. Previous tests determined that there was no change in the lambda max of lornoxicam due to the presence of carrier dissolved in the dissolution medium.²² The dissolution studies were carried out in triplicate and the mean \pm SD values were calculated. The solid dispersion of lornoxicam- β -CD inclusion complex showed maximum 99.67 \pm 0.75% drug release in 90 min than physical mixture 41.10 \pm 0.50% drug release in 120 min and Pure drug lornoxicam only 37.50 \pm 0.82% release in 120 min. (Table 5 & Fig. 9)

Result and Conclusion

Evaluation of Solid dispersions

Evaluation of Lornoxicam with β -CD inclusion complex prepared by kneading method selected on the basis of maximum solubility observed as compared to other method and was evaluated by different evaluation parameters like production yield, drug content, solubility determination Taste evaluation, Differential scanning calorimetry, X-ray diffraction study, FTIR study, Micromeritic properties, SEM and In vitro drug release study.

Saturation solubility studies

Saturation solubility studies were performed according to the method reported by Higuchi and Connors. Lornoxicam with β -CD inclusion complex with ratio 1:2 prepared by kneading method, spray drying method and complexation method. The samples were withdrawn and filtered through a 0.22 µm membrane filter (Millipore, India) followed by dilution and analyzed lornoxicam content was determined in triplicate by (UV/Vis Spectrophotometer, Shimadzu-1800) spectrophotometrically λmax 377.5 nm. at at Lornoxicam with β -CD inclusion complex with ratio (ratio 1:2) shows maximum solubility in 35.34 ± 1.0 mg/ml, 26.24 ± 1.2 mg/ml and 20.19 ± 1.1 mg/ml was found and optimized the kneading method on the basis of maximum solubility and further study was carried out. The results were shown in (Table 2 & Fig.1)

Production Yield

The production yields of Lornoxicam with β -CD inclusion complex with ratio 1:2 prepared by kneading method, spray drying method and complexation method were calculated using the weight of final product after drying with respect to the initial total weight of the drug and polymer used for preparation of solid dispersions and percent production yields were calculated as per the formula. The results are reported in (Table 3 & Fig.2)

Drug content:

Drug content of Lornoxicam with β -CD inclusion complex with ratio 1:2 prepared by kneading method, spray drying method and complexation method, solid dispersion complex weigh equivalent to 10 mg of drug according to theoretical mass was dissolved in the sufficient amount of methanol into 100 mL volumetric flask and volume was made up to the mark by methanol. The samples were withdrawn and filtered through a 0.22 µm membrane filter (Millipore, India)

followed by dilution and analyzed lornoxicam content was determined in triplicate by (UV/Vis Spectrophotometer, Shimadzu-1800) spectrophotometrically at λ max 377.5 nm. Results are reported in (Table 3 & Fig.3)

Taste Evaluation of Lornoxicam with β -CD inclusion complex

Taste evaluation of lornoxicam with β -CD inclusion complex with ratio 1:2 prepared by kneading method was done to compare bitterness of the optimized solid dispersion inclusion complex with pure drugs. Evaluation of taste of drug- β -CD inclusion complex was carried out using time sensitivity method and by using panel of human volunteers. Form the observation most of the volunteers passes the taste of lornoxicam with β -CD inclusion complex with ratio 1:2 which was prepared by kneading method. The results are revealed in (Table 4) Taste scale was calibrated as follows:

- 0 Being = Tasteless
- X Being= Thresholds bitter
- 1 Being= Slight bitter
- 2 Being= Moderate bitter
- 3 Being= Strong bitter

Characterization of Solid dispersions Fourier Transform Infrared Spectroscopy

The IR spectra of the LXM, β -CD, physical mixtures and lornoxicam-β-CD inclusion complex are shown in (Fig.4). The IR spectroscopy has also been used to assess the interaction between cyclodextrin and guest molecules in the solid state, since upon the complexation, shifts or changes in the absorption spectrum occur. The IR spectrum of the LXM showed an absorption band at 3,400 cm-1 due to an O-H stretching vibration; the broadness of this band is indicative of hydrogen bonding. The absorption band at 3,090 cm-1 due to N-H stretching; aromatic C-H stretching at 2,927 cm-1 and an absorption band located at 766 cm-1 is due to the stretching vibration of C –Cl38. The IR spectra of the β -CD had an intense bands at 3,300-3,500 cm-1 due to O-H stretching vibrations and vibration band located at 2,800-3,000 cm-1 due to the -CH and CH2 groups. The IR spectra of lornoxicam-\beta-CD inclusion complex was compared with physical mixtures and LXM, there was no significant change in the characteristics stretching band of LXM were observed in physical mixtures, the above bands are unchanged for position and intensity with respect to the IR spectra of LXM alone in physical mixture. The O-H band of LXM and LXM- β-CD which confirmed the existence of interaction of the drug, β -CD and these spectral changes due to dissociation of intermolecular hydrogen bonds of pure drug through inclusion complexation, similarly the C- O stretching band was highly diminished and interact with β -CD and shifted to lower frequencies in all spectral patterns of solid complexes prepared by kneading method. It was clearly

verified that the magnitude of alteration of the original C-O stretching band was clearly influenced by the method of preparation. This result confirmed the existence of strong interactions between LXM, β -CD. In IR spectra of lornoxicam- β -CD inclusion complex, absorption band at 1078 cm-1 could not be detected which might be due to co- occurrence of C-O band with other band and this existence indicated a strong interaction and complete formation of the solid complex with β -CD. Inclusion complexes showed absence of the above-mentioned peaks, which indicates entrapment of LXM into the cavity and confirmed the complex formation.

Differential Scanning Calorimetry

The thermal behavior of LXM, β -CD, physical mixtures and lornoxicam-\beta-CD inclusion complex is shown in (Fig.5). The DSC thermograms of LXM showed a typical behavior of an anhydrous crystalline drug with a well defined endothermic peak at 230°C corresponding to its melting point in optimized solid complex. The DSC thermogram of β -CD showed an endothermic peak at about 135°C due to the liberation of water of crystalline, Both the characteristic peak of LXM and β -CD was clearly distinguishable in the physical mixture.. The DSC curves of lornoxicam-\beta-CD inclusion complex exhibited a complete disappearance of the endothermic melting peak of LXM and this may be attributed to amorphization of LXM and due to formation of inclusion complex. Disappearance of endothermic melting peak of LXM showed that only lornoxicam- β -CD inclusion complex formed a true inclusion complex which was differing from physical mixture.

X-Ray Powder Diffractometry

The XRPD patterns of pure LXM, β -CD, physical mixture and lornoxicam- β -CD inclusion complex is shown in (Fig.6). It is a useful method for the detection of complexation in powder of microcrystalline states. The diffractogram of LXM showed sharp and intense peaks which indicating its crystalline nature. The XRPD pattern of pure β -CD showed crystallinity but in the comparison with the diffractogram of the correspondent to the physical mixture, it was possible to observe disappearance of some diffraction peaks of β -CD and LXM. Furthermore, for lornoxicam- β -CD inclusion complex obtained XRPD patterns were indicating the amorphous state reached by kneading method. SEM and DSC studies also supported the same hypothesis, which was confirmed by X-ray diffractometry.

Scanning Electron Microscopy

Scanning electron microscopy photographs of optimized Lornoxicam and batches of ratio 1:2 lornoxicam- β -CD inclusion complex are given in (Fig.7 & 8). SEM study indicated that pure drug Lornoxicam in particles were irregular in shape and crystalline in nature, while inclusion complex of drug and carrier shows that drug particle remains

http://www.ijddhrjournal.com.

dispersed and physically adsorbed on the surface of the carrier particles. The solid dispersion of lornoxicam- β -CD inclusion complex showed a homogeneous dispersions indicating that the Lornoxicam molecules were dispersed uniformly in carriers matrices of solid dispersions prepared kneading method. It can be seen in (Fig.5 & 6) which shows crystalline nature of drug convert into amorphous state, which is further confirmed by DSC and XRD study.

In-vitro dissolution study:

The in-vitro dissolution of lornoxicam, physical mixture and prepared solid inclusion complexes with β -CD and was carried out using powder dispersion method by using Dissolution test apparatus-TDT-06T (Electrolab, Mumbai, India) at the USP type II apparatus at 100 rpm. The dissolution study was conducted in dissolution media at $37^{\circ}C \pm 0.5^{\circ}C$. Optimized batches of ratio 1:2 lornoxicam- β -CD inclusion complex containing equivalent of 10mg of lornoxicam were suspended in 900 ml of pH 6.8 and a 5 ml aliquot of dissolution medium was withdrawn at 5, 10, 15, 30, 45, 60, 90, 105, 120 min with a pipette and filter through 0.45 µm Whatman filter and then analyzed lornoxicam content was determined in triplicate by (UV/Vis Spectrophotometer, Shimadzu-1800) spectrophotometrically at λmax 377.5 nm. The statistical analysis of the data is done using Microsoft Excel[®] and the data are presented in (Table 17 & Fig. 33). The solid dispersion of lornoxicam-\beta-CD inclusion complex showed maximum 99.67 \pm 0.75% drug release in 90 min than physical mixture $41.10 \pm 0.50\%$ drug release in 120 min and Pure drug lornoxicam only $37.50 \pm 0.82\%$ release in 120 min.

Conclusion:

Lornoxicam is a non-steroidal anti-inflammatory drug with extremely potent anti-inflammatory and analgesic activity. But it shows bitter taste and distinct pHdependent solubility characterized by very poor solubility in acidic condition present in the stomach. The research work was start with aim of preparation and evaluation of lornoxicam-\beta-CD inclusion complex of solid dispersion were prepared by kneading method, spray drying method and complexation method in different ratio like 1:1, 1:2, 1:3 & 1:4 and optimized the ratio on the basis of maximum solubility of drug in prepared complex at pH 1.2. The ratio (ratio 1:2) shows maximum solubility in 35.34 ± 1.0 mg/ml which was selected for further study. like production yield 95.12%, drug content 99.84±1.6%, solubility 35.34 ± 1.0 mg/ml, passess taste with scale i.e. 0 Being = Tasteless of lornoxicam with β -CD inclusion complex, Solid dispersions of lornoxicam with β-CD inclusion complex was characterized and confirmed by Xray diffractometry, SEM and DSC, FTIR studies also supported the same hypothesis. The dissolution study was conducted in dissolution media having pH 6.8 as a dissolution media at $37^{\circ}C \pm 0.5^{\circ}C$. The solid dispersion of lornoxicam- β -CD inclusion complex showed maximum 99.67 \pm 0.75% drug release in 90 min than physical mixture 41.10 \pm 0.50% drug release in 120 min and Pure drug lornoxicam only 37.50 \pm 0.82% release in 120 min. It can be concluded that inclusion complex of lornoxicam with β -CD was prepared by kneading method used for taste masking of bitter drug (lornoxicam) and solubility enhancement were found to be effective for solubility and dissolution rate, which leads to enhanced bioavailability of lornoxicam.

References

- Abdou, H.M., 1989. Dissolution, Bioavailability and Bioequivalence. Theory of dissolution, Ch. 2. Mack, PA, p. 11.
- Otsuka, M., Kaneniwa, N., 1984. Effects of grinding on the physicochemical properties of cephalexin. Chem. Pharm. Bull. 32, 1071–1079.
- 3) Ford, J.L., 1986. The current status of solid dispersions. Pharm. Acta Helv. 61, 69–88.
- Takeuchi, H., Handa, T., Kawashima, Y., 1987. Enhancement of the dissolution rate of a poorly water-soluble drug (tolbutamide) by a spray-drying solvent deposition method and disintegrants. J. Pharm. Pharmacol. 39, 769–773.
- Nystrom, C., Westerberg, M., 1985. The use of ordered mixtures for improving the dissolution rate of low solubility compounds. J. Pharm. Pharmacol. 38, 161–165.
- Nozawa, Y., Mizumoto, T., Higashide, F., 1984. Increasing dissolution rate of phenacetin by roll mixing with polyvinyl pyrrolidone. Yakuzaigaku 44, 134–140.
- Moyano, J.R., Gines, J.M., Arias, M.J., Rabasco, A.M., 1995. Study of dissolution characteristics of oxazepam via Complexation with β-cyclodextrin. Int. J. Pharm. 114, 95–102.
- 8) Masaaki Sugimoto, Takuya Okagaki, Shinji Narisawa, Yoshiyuki Koida, Kingo Nakajima. Improvement of dissolution characteristics and bioavailability of poorly water-soluble drugs by novel cogrinding method using water-soluble polymer. . Int. J. Pharm. 160(1998) 11-19.
- Marcı'lio S.S. Cunha-Filho, Ramo'n Martı'nez-Pacheco, Mariana Landı'n. Dissolution rate enhancement of the novel antitumoral β-lapachone by solvent change precipitation of microparticles. Eur. J. Pharm. Biopharm. 69 (2008) 871–877.
- D. Douroumis, A. Fahr. Nano- and microparticulate formulations of poorly water-soluble drugs by using a novel optimized technique. Eur. J. Pharm. Biopharm. 63 (2006) 173–175.
- P. Kocbek, S. Baumgartner, J. Kristl. Preparation and evaluation of nanosuspensions for enhancing the dissolution of poorly soluble drugs. Int. J. Pharm. 312 (2006) 179–186.

- 12) Yousef Javadzadeh, Baharak Jafari-Navimipour, Ali Nokhodchi. Liquisolid technique for dissolution rate enhancement of a high dose water-insoluble drug (carbamazepine). . Int. J. Pharm. 341 (2007) 26–34.
- 13) Michael Lee Branham, Thomas Moyo, Thirumala Govender. Preparation and solid-state characterization of ball milled saquinavir mesylate for solubility enhancement. Eur. J. Pharm. Biopharm. 80 (2012) 194–202.
- 14) Mitra Mosharraf, Christer Nystrom. The effect of dry mixing on the apparent solubility of hydrophobic, sparingly soluble drugs. Eur. J. Pharm. Sci. 9 (1999) 145–156.
- 15) Sumio Chono, Eri Takeda, Toshinobu Seki, Kazuhiro Morimoto. Enhancement of the dissolution rate and gastrointestinal absorption of pranlukast as a model poorly water-soluble drug by grinding with gelatin. . Int. J. Pharm. 347 (2008) 71–78.
- Douroumis, A. Fahr. Stable carbamazepine colloidal systems using the cosolvent technique. Eur. J. Pharm. Sci. 30 (2007) 367–374.
- 17) Kalakuntla, D., Gada, M., Kulkarni, G., 2011. Design And Development Of Taste Masking Lornoxicam Orodispersible Tablet, Ijpi's Journal Of Pharmaceutics And Cosmetology Vol 1(2), 121-125.
- 18) Scalia, S., Villani, S., Scatturin, A., Vandelli, M.A., Forni, F., 1998. Complexation of the sunscreen agent, butylmethoxydibenzoylmethane, with hydroxypropyl-β-cyclodextrin. Int. J. Pharm. 175, 205–213.
- Mielcarek, J., Daczkowska, E., 1999. Photodegradation of inclusion complexes of isradipine with methyl-β-cyclodextrin. J. Pharm. Biomed. Anal. 21, 393–398.
- 20) Sortino, S., Scaiano, J.C., De Guidi, G., Monti, S., 1999. Effect of β-cyclodextrin complexation on the photochemical and photosensitizing properties of tolmetin: a steady state and time-resolved study. Photochem. Photobiol. 70, 549–556.
- 21) Fareen Sami, Betty Philip, and Kamla Pathak, Effect of Auxiliary Substances on Complexation Efficiency and Intrinsic Dissolution Rate of Gemfibrozil–β-CD Complexes. AAPS PharmSciTech. 2010 March; 11(1): 27–35
- 22) Fernandes C. M., Vieira M. T., and Veiga F. J. B., Physicochemical characterization and in vitro dissolution behavior of nicardipine– β- cyclodextrins inclusion compounds, Eur. J. Pharm. Sci., 2002, 15, 79–88.
- 23) Kamal Dua, Kavita Pabreja, M. V. Ramana, and Vinny Lather. Dissolution behavior of β-cyclodextrin molecular inclusion complexes of aceclofenac, J Pharm Bioallied Sci. 2011 Jul-Sep; 3(3): 417–425.

- 24) Higuchi T and Connors KA. Advances In Analytical Chemistry Instrumentation. Wiley-Interscience, New York (1965) 117-212.
- 25) Connors KA. A Textbook of Pharmaceutical Analysis. 3rd ed., Wiley-Interscience, New York (1982) 328-339.
- 26) Kulkarni Ajit Shankarrao, Ghadge Dhairysheel Mahadeo and Kokate Pankaj Balavantrao, Formulation and *In-vitro* Evaluation of Orally Disintegrating Tablets of Olanzapine-2-Hydroxypropyl-β-Cyclodextrin Inclusion Complex. Iranian Journal of Pharmaceutical Research (2010), 9 (4): 335-347.
- 27) Borodkin, S. Ion Exchange Resin Delivery system, In "Polymers for controlled Drug Delivery (P.J. Tarcha, ed.) CRC Press, Inc, Boca Raton, PP 215-230, 1991.

Table 1: Ratio optimization of drug and β-Cyclodextrin

Ratio	Drug Concentration (Solubility in mg/ml)
1:1	1.6 ± 0.7
1:2	2.6 ± 0.8
1:3	1.9 ± 0.4
1:4	2.2 ± 0.7

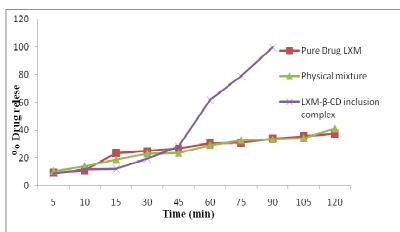


Fig. 9: In-vitro dissolution profile of LXM, Physical mixture and LXM-β-CD inclusion complex

http://www.ijddhrjournal.com.

F F F F F F F F F F		
Methods	Solubility ± SD (mg/ml)	
Pure LXM	0.034 ± 1.0	
Kneading Method	35.34 ± 1.0	
Spray drying method	26.24 ± 1.2	
Complexation method	20.19 ± 1.1	

Table 2: Solubility study of solid dis	spersion inclusion (complexes at pH 1.2
Tuble 2. Bolubility study of solid dis	persion merusion	complexes at pit 1.2

Table 3: Evaluation of Lornoxicam with β -CD inclusion complex

Lornoxicam with β-CD inclusion complex (1:2 ration)	% Yield	% Drug content	Solubility mg/ml (Mean ± SD)
Kneading Method	95.12	99.84±1.6	35.34 ± 1.0
Spray drying method	85.67	88.45±0.64	26.24 ± 1.2
Complexation method	93.95	90.02±0.74	20.19 ± 1.1

Table 4: Taste Evaluation of Lornoxicam with β -CD inclusion complex

T 7 L 4	Bitterness level after					
Volunteers	10 sec.	1 min.	2 min.	5 min.	10 min.	15min.
1	X	0	0	0	0	0
2	X	X	0	0	0	0
3	0	0	0	0	0	0
4	0	0	0	0	0	0
5	0	0	0	0	0	0
6	0	0	0	0	0	0
7	X	0	0	0	0	0
8	0	0	0	0	0	0
9	0	0	0	0	0	0
10	0	0	0	0	0	0

Time	Pure Drug LXM	Physical mixture	LXM-β-CD inclusion complex
5	9.21 ± 0.67	10.47 ± 0.62	8.64 ± 0.35
10	10.94 ± 1.28	14.17 ± 0.93	11.44 ± 0.64
15	23.48 ± 1.07	18.73 ± 0.77	11.97 ± 0.94
30	24.87 ± 0.89	23.18 ± 0.33	19.24 ± 0.77
45	26.69 ± 0.68	23.59 ± 0.60	27.94 ± 0.68
60	30.48 ± 1.13	29.05 ± 0.67	61.71 ± 0.92
75	30.92 ± 1.09	32.71 ± 0.67	78.80 ± 0.99
90	33.90 ± 0.86	33.31 ± 0.76	99.67 ± 0.75
105	35.76 ± 0.83	34.05 ± 0.71	
120	37.50 ± 0.82	41.10 ± 0.50	

Table 5. In-vitro dissolution study	of LXM. Physical mixture and LXM-B-CD	inclusion complex
Table 3. III-villo uissolulloli sluu		пилизии сотприх

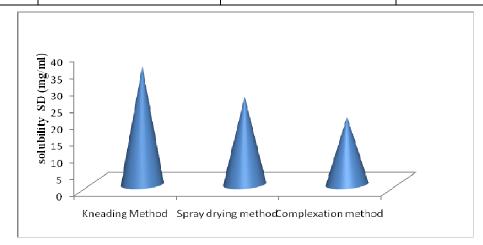
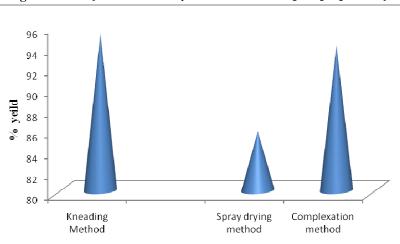


Fig.1: Solubility studies LXM-β-CD inclusion complex prepared by method



(C)Int. J. of Drug Discovery & Herbal Research

Fig.2: Production Yield of Lornoxicam with β-CD inclusion complex prepared by method

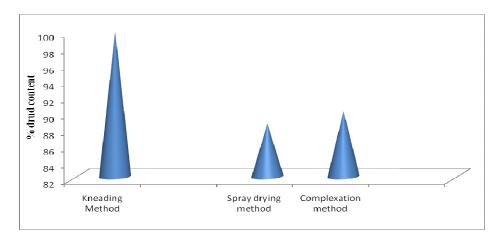


Fig.3: Drug Content of Lornoxicam with β-CD inclusion complex prepared by method

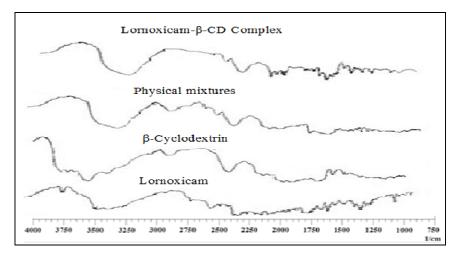
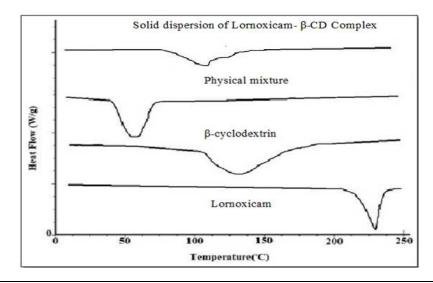


Fig. 4: FTIR spectra of Lornoxicam, β-Cyclodextrin, Physical mixture and Lornoxicam-β-CD inclusion complex



(C)Int. J. of Drug Discovery & Herbal Research 5.



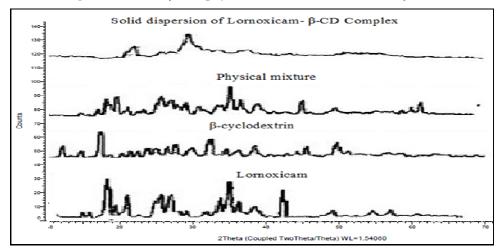


Fig. 6: Diffractogram of LXM, β-CD, physical mixtures and lornoxicam-β-CD inclusion complex

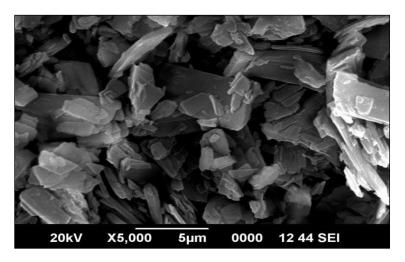


Fig. 7: SEM of Lornoxicam

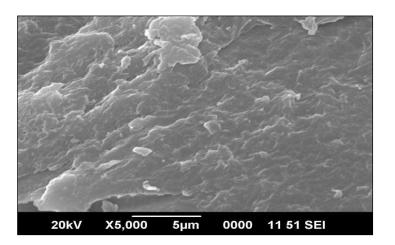


Fig. 8: SEM of solid dispersion of lornoxicam-β-CD inclusion complex

http://www.ijddhrjournal.com.

(C)Int. J. of Drug Discovery & Herbal Research